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1s 2s 2p 6 3s 2 3p 3 d 10 4s 1 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 9. 1 1012 m What is the wavelength of light that ...

~~4 Arrangement of Electrons in Atoms~~

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Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. ____ 7. A solution containing 35 g of Li 2SO 4 dissolved in 100 g of water is heated from 10 ° C to 90 ° C. According to information in the figure, this temperature change would result in a. an additional 5 g of Li 2SO 4 in ...

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Modern Chemistry 27 Chapter Test Name Class Date Chapter Test A, continued ____ 4. The electron configuration below violates a. the Pauli exclusion principle. b. the Aufbau principle. c. Hund ' s rule. d. Both (a) and (c) ____ 5. A photon is emitted from a gaseous atom when an electron moves to its ground state from a(n) a. inner shell. b ...

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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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