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T2 = final temperature (K). Chemistry: Form WS6.4.1A. Name ... 4. 4.5 L. 1.035 atm. 375 K. 1.100 atm. 350. K. 5. 0.980 atm. 30.0 ° C. 185 mL. 0.900 atm. 28.0 ° C. 6 . WS6-4-1A\_Using\_the\_Combined\_Gas\_Law.pdf

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Chemistry: Form N6.4.1A Using the Combined Gas Law PHASES OF MATTER Page 2 © Evan P. Silberstein, 2003 Fill in the blanks in the table below by using the combined gas law to find the unknown variable in each row. Initial Final volume pressure temperature volume pressure temperature 1. 80.0 mL 96 kPa 27 ° C 99.0 kPa 0.0 ° C

Using the Combined Gas Law - Evan's Regents Chemistry Corner

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Chem 1a Exam 1

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