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Chapter 6 Answers Page 195 #1. Use the tables on page 420 (heat of dissolution)

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and 422 (polyatomic ions) to help you answer these questions. a) $\text{HClO}_3 (\text{s}) \rightarrow \text{H}^+ (\text{aq}) + \text{ClO}_3^- (\text{aq})$ $\Delta H =$ b) $\text{HI} (\text{s}) \rightarrow \text{H}^+ (\text{aq}) + \text{I}^- (\text{aq})$ $\Delta H =$ c) $\text{KNO}_3 (\text{s}) \rightarrow \text{K}^+ (\text{aq}) + \text{NO}_3^- (\text{aq})$... <https://studylib.net/doc/5610530/chemistry-chapter-6-answer-key>.

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Modern Chemistry 47 Chemical Bonding
CHAPTER 6 REVIEW Chemical Bonding
SECTION 4 SHORT ANSWER Answer
the following questions in the space

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provided. 1. _____ In metals, the valence electrons are considered to be (a) attached to particular positive ions.

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CHAPTER 6 REVIEW Chemical Bonding
SECTION 1 SHORT ANSWER Answer

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the following questions in the space provided.

1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures

2. b A covalent bond consists of (a) a shared electron.

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~~6 Chemical Bonding Effingham County Schools / Overview~~

SECTION 6.1 Nature favors arrangements in which potential energy is minimized. For example, a boulder is less likely to balance at the top of a hill than it is to roll to the bottom of a valley. The boulder at

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the top of the hill is not stable. Atoms are usually not stable when isolated. They usually combine to form more stable arrangements of matter.

~~CHAPTER 6 Chemical Bonding~~

molar mass of $C_6H_6O = 94.11 \text{ g}$ 21. a.

molar mass of $SO_3 = 80.07 \text{ g}$ 49.2 mg

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SO₃ 1 g 1000 mg 1 mol 80.07 g = 6.14
10⁻⁴ mol SO₃ 3 b. molar mass of PbO₂ =
239.2 g 7.44 x 10⁴ kg PbO₂ 1000 g 1 kg
1 mol 239.2 g = 3.11 10⁵ mol PbO₂ c.
molar mass of CHCl₃ = 119.37 g 59.1 g
CHCl₃ 1 mol 119.37 g = 0.495 mol
CHCl₃ d. molar mass of C₂H₃Cl₃ =
133.39 g 3.27 ...

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Chemical Bonding SECTION 1 SHORT
ANSWER Answer the following questions
in the space provided. 1. a A chemical

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bond between atoms results from the attraction between the valence electrons and of different atoms.

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Organic Chemistry Chapter 6 and 7

questionalkyl halide answerhalogen atom

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bonded to an sp^3 hybridized carbon atom
that is tetrahedral question primary alkyl
halide answer One carbon attached ... Test
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