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Hydrogen Ions and pH

pages 650 – 658 Practice

Problems pages

651 – 657 22. The

concentration of either

the H^+ ion or the OH^-

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ion is given for four
aqueous solutions at 298
K. For each solution,
calculate $[H^+]$ or
 $[OH^-]$. State whether the
solution is acidic,

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Electrochemistry

MULTIPLE CHOICE

QUESTIONS NOTE: A

table of standard

reduction potentials is

required to work many of

these problems. 1.

Complete and balance

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the following redox equation. When properly balanced using the smallest whole-number coefficients, the coefficient of S is $\text{H}_2\text{S} + \text{HNO}_3 \rightarrow \text{S} + \text{NO}$ (acidic solution) 2.

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