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$3 \times 10^5 \text{ Pa} = \frac{P_{\text{vap}}}{P} \times 18 \text{ mol} \times 8.314 \text{ J/mol}\cdot\text{K} \times T$

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10<sup>5</sup> (bar) m<sup>3</sup> mol K ) T

$P = \frac{V}{V} \times 10^3 \times (10^5 \times x)$

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