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a) O atom = 0, C atom = 0, and N atom = 0 b) O atom = 0, C atom = 0,

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and N atom = - c) O atom = -1, C atom = 0, and N atom = 0 d) O atom = -1, C atom = -1, and N atom = + e) O atom = +1, C atom = 0, and N atom = -. 17. Use VSEPR to predict the electron-pair geometry and the molecular geometry of sulfur dioxide, SO<sub>2</sub>.

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Identify the letter of the choice that best completes the statement or answers the question and legibly write the letter~~

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