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Created by. kadunning PLUS. 2.1 Properties of Matter 2.2 Mixtures 2.3 Elements and Compounds 2.4 Chemical Reactions. Key Concepts: Terms in this set (121) solution. a homogeneous mixture of two or more substances. gas.

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Section 2.1 Properties of Matter

OBJECTIVES: Describe a physical change. Matter Matter is anything that:

a) has mass, and b) takes up space

Mass = a measure of the amount of

stuff (or material) the object contains

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(don't confuse this with weight, a measure of gravity) Volume = a measure of the space occupied by the object Describing Matter Properties used to describe matter can be classified as: 1) Extensive - depends on the amount of matter in the sample - Mass, volume ...

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~~Chapter 2 \u201cMatter ...~~

Chemistry Chapter 2 Test: Matter and Change 87 Terms. annakoutelas.

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Matter and Change (Chapter 2) mass. volume. extensive property. intensive property. the amount of matter in an object. the amount of space an object takes up. a property that depends on the amount of matter. a property that depends on the type of matter, not the amount.

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Flashcards and Study Sets ...~~

Chapter 2 - Matter and Change. 1. Chapter 2 □ Matter & Change By Kendon Smith □ Columbia Central HS □ Brooklyn, MI. 2. Chapter 2 □ Matter and Change. Properties of Matter 1. Matter is defined as anything that has mass and volume. a. Mass: the amount of matter an object contains.

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SECTION 2.3 ELEMENTS AND COMPOUNDS (pages 48–52) This section explains a key difference between an element and a compound, and describes how chemical symbols

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and formulas are used to represent elements and compounds. It also summarizes the process for classifying substances and mixtures.

~~SECTION 2.1 PROPERTIES OF MATTER (pages 39-42)~~

MATTER AND CHANGE. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. mrgarner. From chapter 2 of Prentice Hall Chemistry. Key Concepts: Terms in this set (32) mass. the amount of matter in an object. volume. the amount of space occupied by an object. extensive. properties of matter like volume, mass, and amount of ...

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The subscript 2 in H_2O indicates that there are always two parts of

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hydrogen for each part of oxygen in water. Because a compound has a fixed composition, the formula for a compound is always the same.

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MATTER Matter is anything that: a) has mass, and b) takes up space
Mass = a measure of the amount of

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Chapter 2 □ Matter and Change

Chapter 2: 1 - 12, 13, 14 □ 23, 29, 33, 38, 40, 43, 45 (26 total) Section

Review 2.1 1. Is every sample of matter a substance? Explain. 2.

Contrast the characteristics of the three states of matter. a. b. c. 3. Which of the following are physical changes?

a. making caramel from sugar b. carving a wooden figurine

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Properties of Matter - 2.1 Lesson
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Matter and Change—2 ...~~

What kind of change in matter involves
a change in state, change in size, or a
change in shape? Chapter 2 Matter
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~~CHAPTER 1 REVIEW 23 MATTER AND CHANGE 8 Define chemical...~~
CHAPTER 1 REVIEW Matter and Change SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. A horizontal row of elements in the periodic table is called a(n) period . 2. The symbol for the element in Period

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2, Group 13, is B . 3. Elements that are good conductors of heat and electricity are metals . 4.

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Title: Chapter 2 Matter and Change 1 Chapter 2 Matter and Change START IT UP PUP. Unit Notes; 2 Section 2.1 Properties of Matter. What is matter? 3 Which of these is matter? water air light darkness smoke human diamond cold heat sound jello cloud 4 What is matter? Matter is anything that has mass and takes up space. Matter can be classified by ...

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