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2? A + 2B + 3C + 2D + 4Fe. 2. O₃ + xCO . yFe + 3CO. 2. This equation will be balanced when. A . x = 1 and . y = 2 B . x = 2 and . y = 2 C . x = 3 and . y = 2 D . x = 2 and . y = 3. 5. An acidic solution contains A only hydrogen ions B only hydroxide ions C more hydrogen ions than hydroxide ions D more hydroxide ions than hydrogen ions. 6.

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 $2\text{ (g)} \rightarrow 3\text{CO}_2\text{ (g)} + 4\text{H}_2\text{O}(\square)$ Name the type of enthalpy change which the student measured. No marks should be awarded
for 'burning' since the word "burned" appears in the text. (o) Unless the question is clearly about a non-chemistry issue, eg
costs in industrial chemical process, a non-chemical answer gains no marks.

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remaining questions have been asked in the previous set. Question 1. Give one example each of sol and gel. [1] Answer: Sol-
paint, gel-butter. Question 3. Write the IUPAC name of the compound [1] Answer: 3-aminebutan-1-al. Question 5.

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2 by electricity / (electric) current / (flow of) electrons Mark independently b A = chlorine / Cl₂ Ignore Cl 3 B = hydrogen / H₂
Ignore H C = sodium hydroxide / NaOH Ignore references to sodium chloride If both name and formula given, both must be
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